

Interdisciplinary Investigation of the CO₂ Sequestration in Depleted Shale Gas Formations

Project Number DE-FE-0004731

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U.S. Department of Energy
National Energy Technology Laboratory
Carbon Storage R&D Project Review Meeting
Developing the Technologies and Building the
Infrastructure for CO₂ Storage
August 21-23, 2012

Outline

- **Project Benefits**
- Technical Status
 - Imaging at mm- to micron-scales using CT
 - Permeability measurements and application of the Klinkenberg effect
 - Molecular Dynamics simulations for permeability and viscosity estimates
- Accomplishments to Date
- Summary

Benefit to the Program

- Carbon Storage Program major goals
 - Imaging of gas shale from nano to mmCT scanning provides knowledge of the pore connectivity at micro (less than 2nm), meso (2 – 50nm), and macro (> 50 nm) scales
 - Gas permeability measurements taking into account the Klinkenberg effect can lead to pore-size estimates and contributions of different length scales to transport mechanisms
 - Molecular simulation using Monte Carlo and Molecular Dynamics approaches can lead to prediction of density and viscosity estimates at nano and mesoscales

Benefit to the Program

- Project benefits statement
 - The research project is to conduct a multiscale, multiphysics, laboratory study coupled with molecular simulations to assess the feasibility of depleted organic-rich gas shale reservoirs for large-scale CO₂ sequestration. This project supports the Carbon Storage Program's efforts to identify and utilize geological formations capable of storing appreciable volumes of CO₂ with 99% storage permeance in addition to laying the groundwork for realistic estimates of storage capacity of gas shales.

Project Overview: Goals and Objectives

- Determine how physical and chemical processes of CO₂ storage in organic-rich gas shales affect injectivity and storage capacity
- Determine the ability of gas shale to sequester CO₂ (as free vs adsorbed gas) over long periods of time
- Delineate the physical and chemical aspects of CO₂-shale interaction
- Characterize transport processes and mobility of CO₂ in fractures, shale matrix, and pores
- Probe potential interactions of CO₂ with ground water
- Develop a trap and seal framework for CO₂ storage in gas shale reservoirs

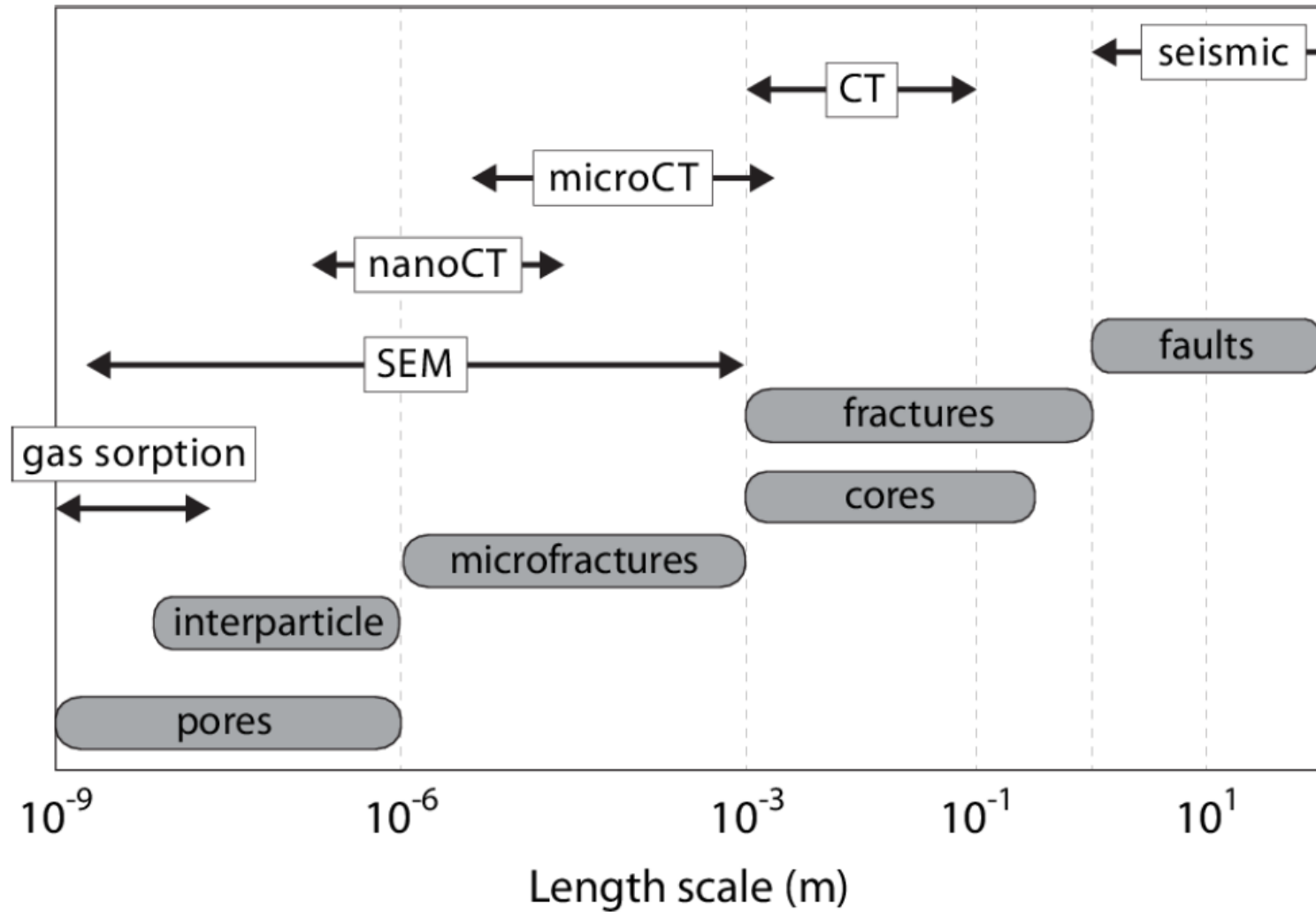
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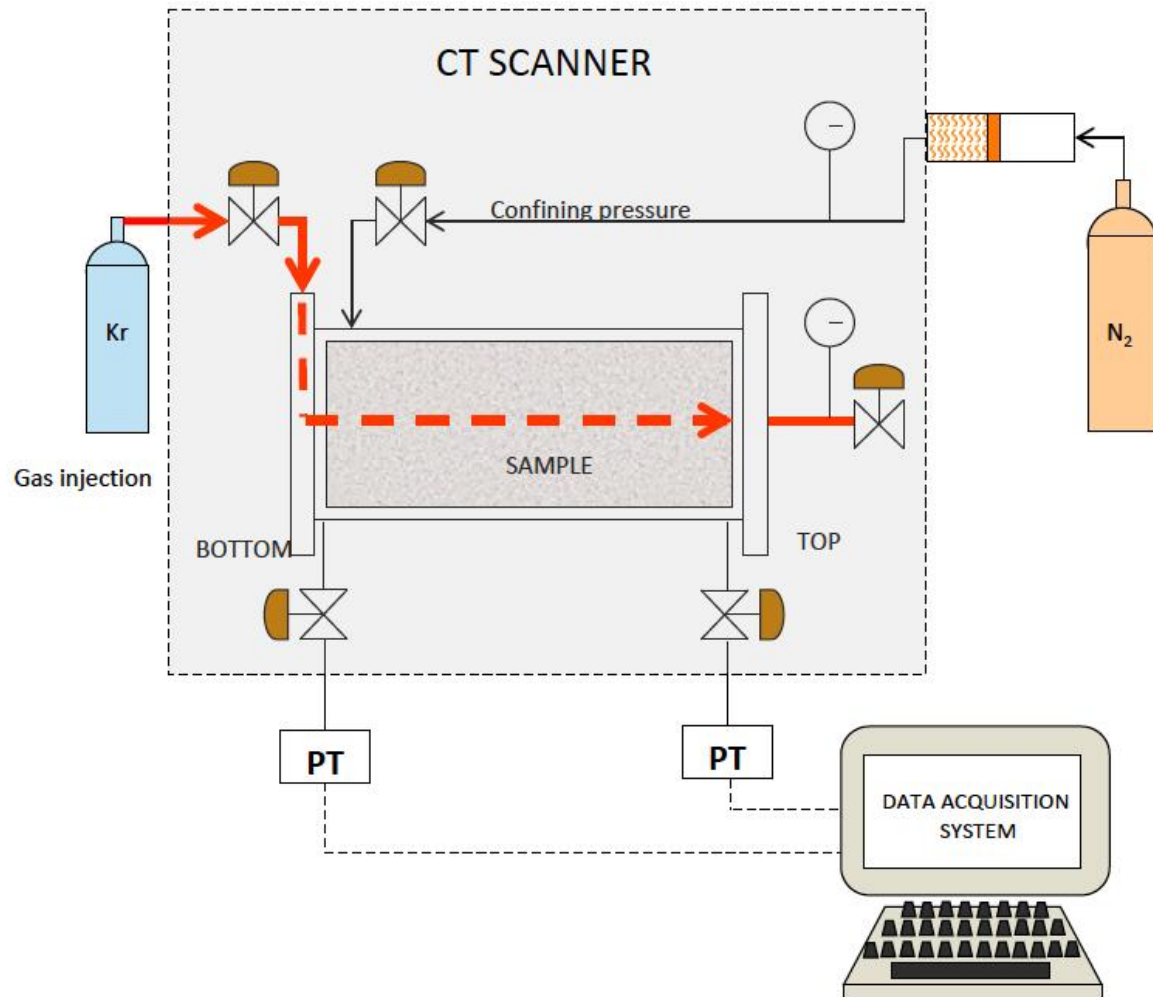
Technical Status

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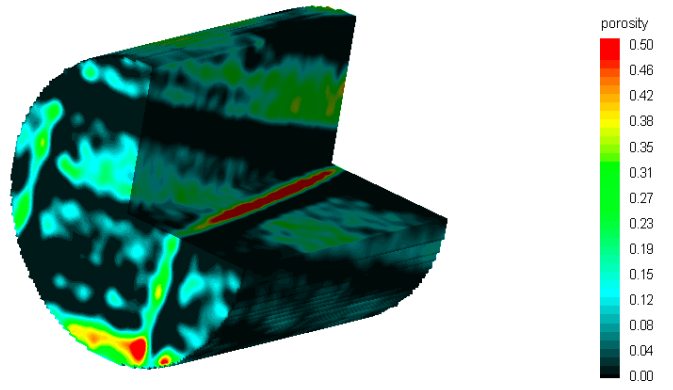
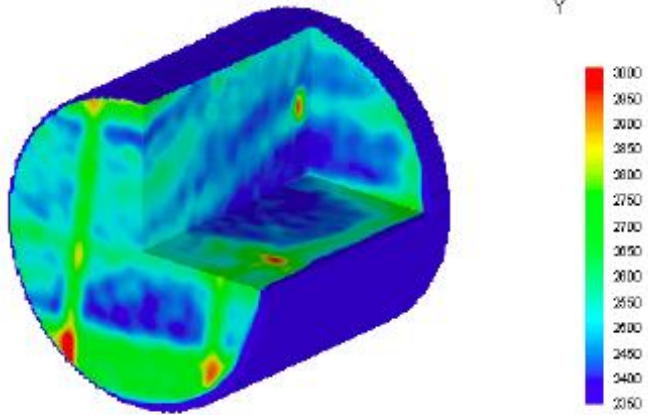
Shale Imaging Tools



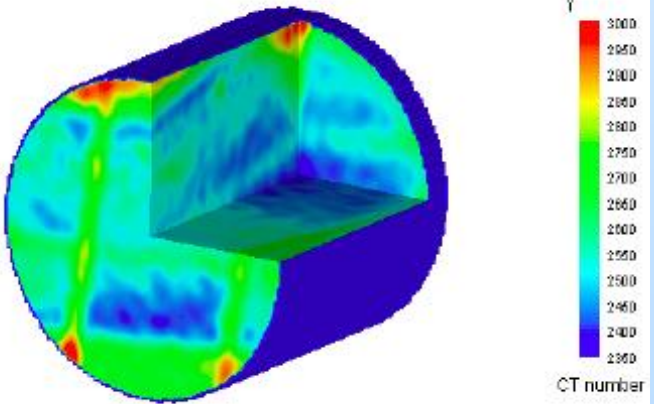
Experimental CT Set up for Kr Flow *mm-scale*



DRY SAMPLE



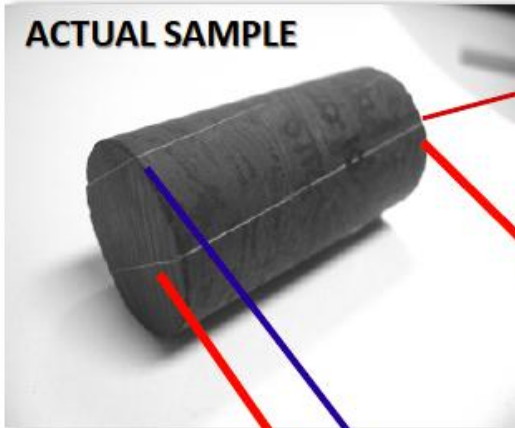
SAMPLE + KR



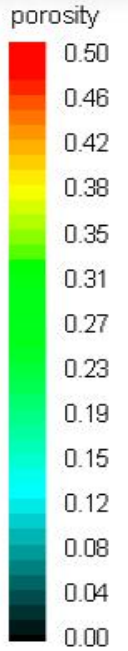
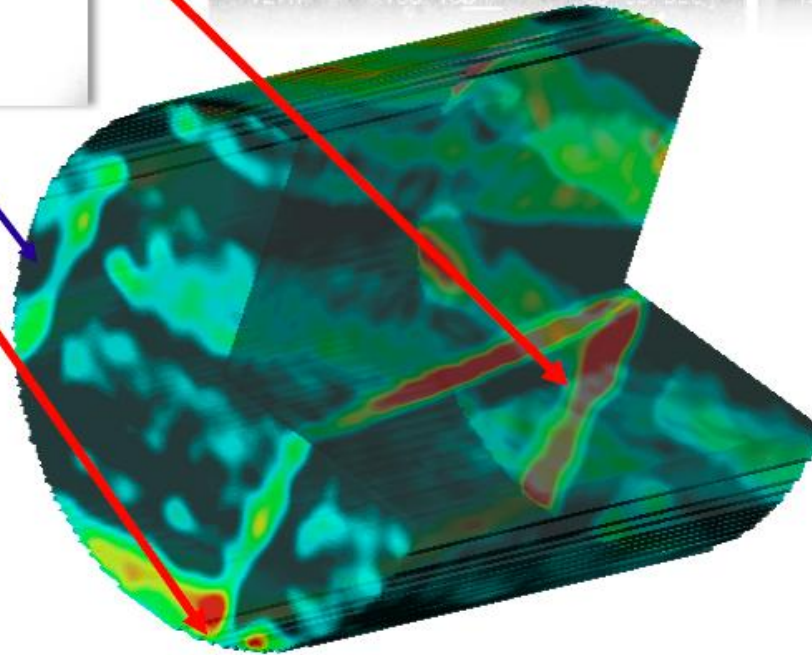
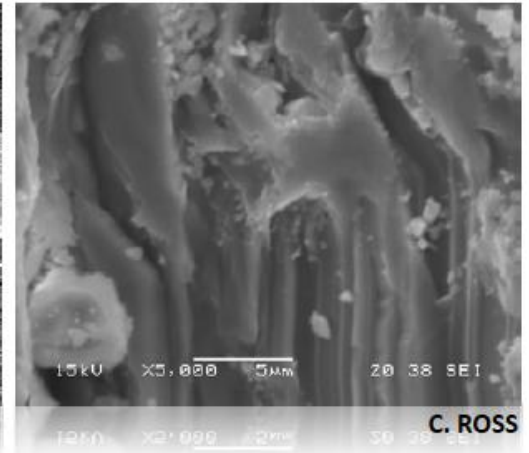
Porosity 3D Image Reconstruction

core length: 51.8 mm
core diameter: 1 in

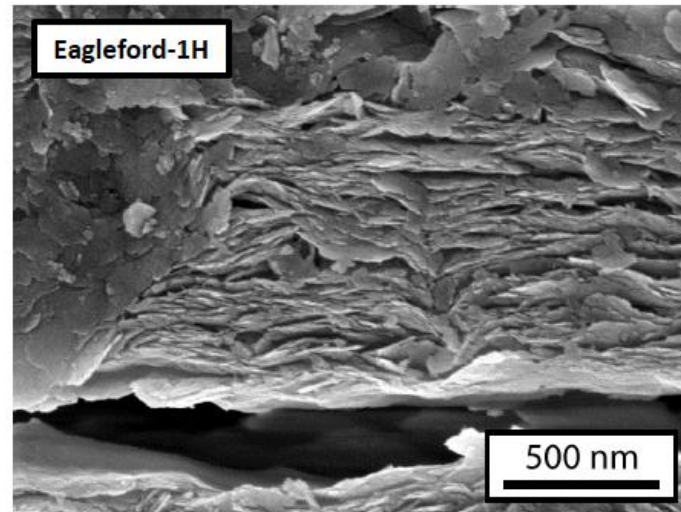
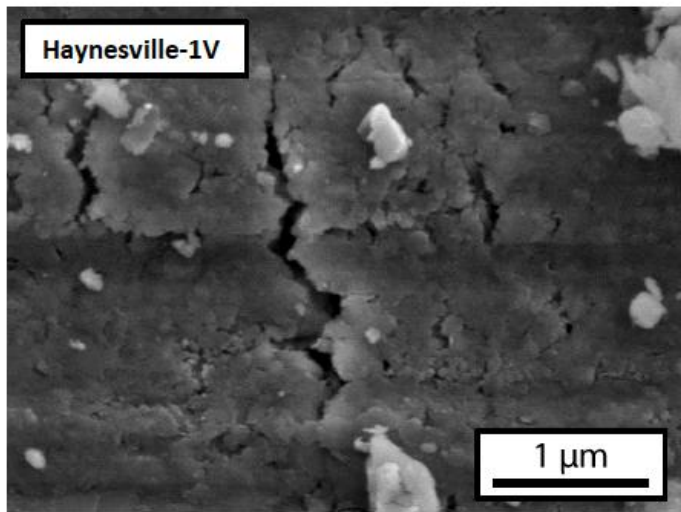
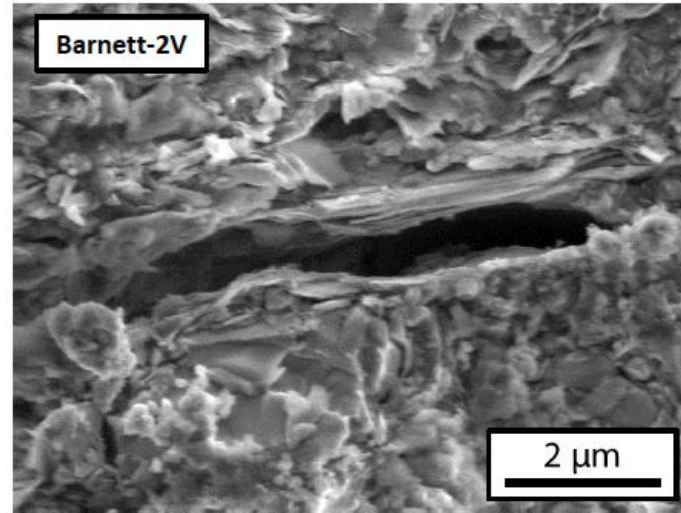
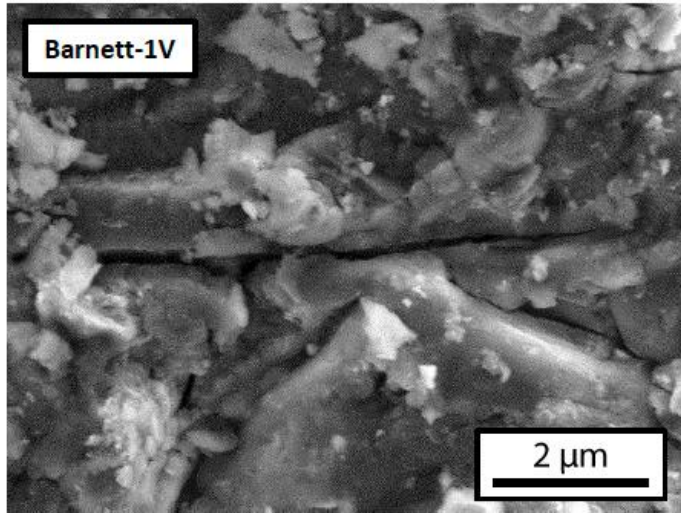
ACTUAL SAMPLE



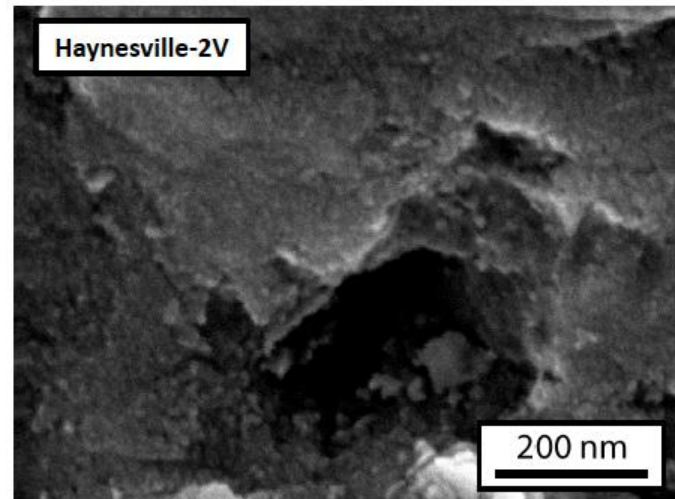
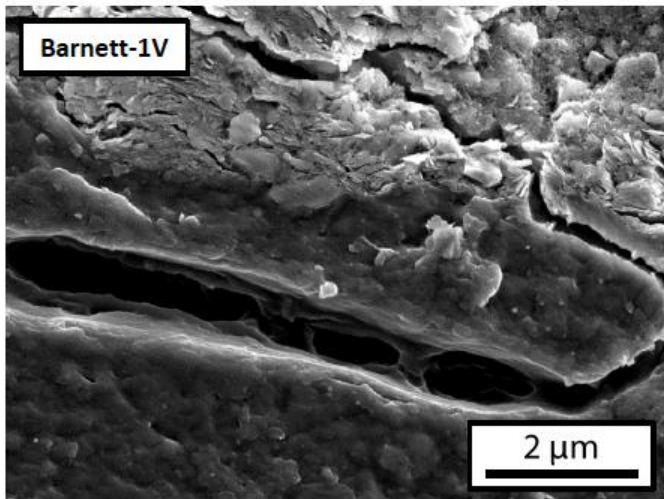
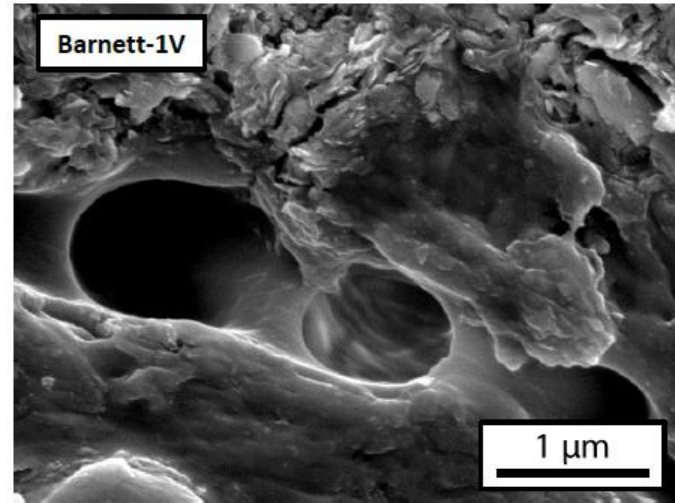
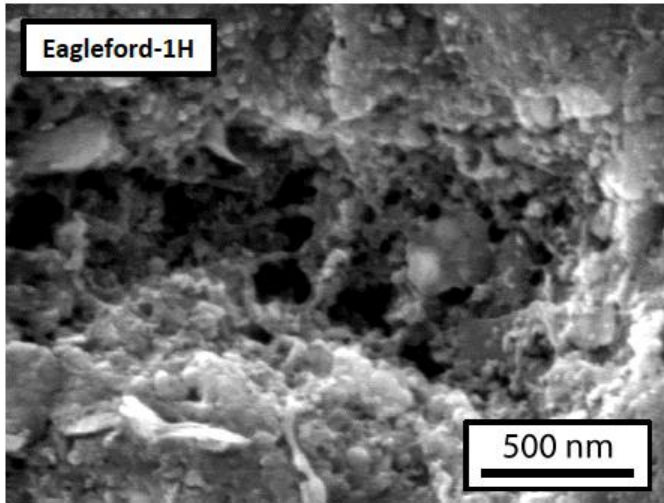
CALCITE-FILLED FRACTURES



Shale Porosity - Microfractures *using microCT*



Shale Porosity – Organic Matter *using microCT*



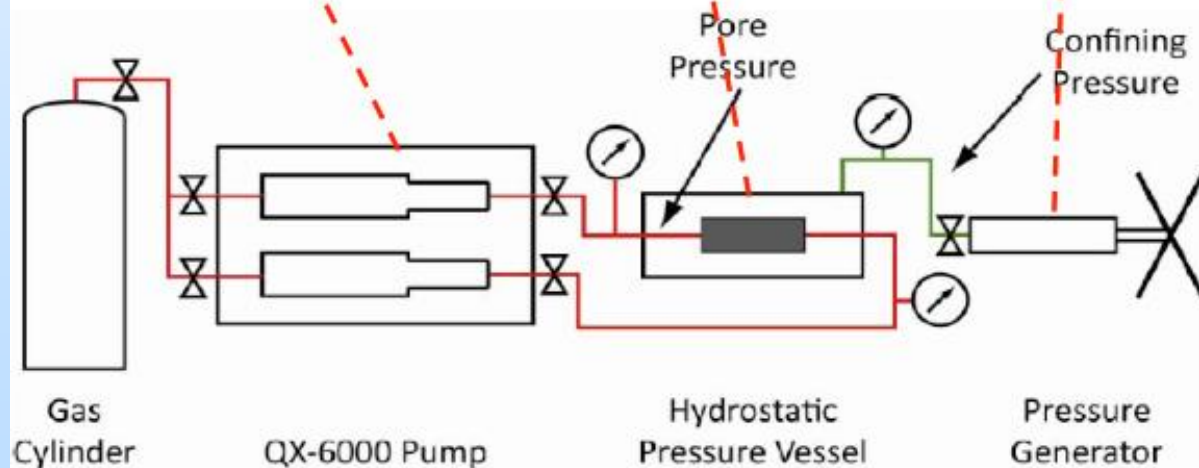
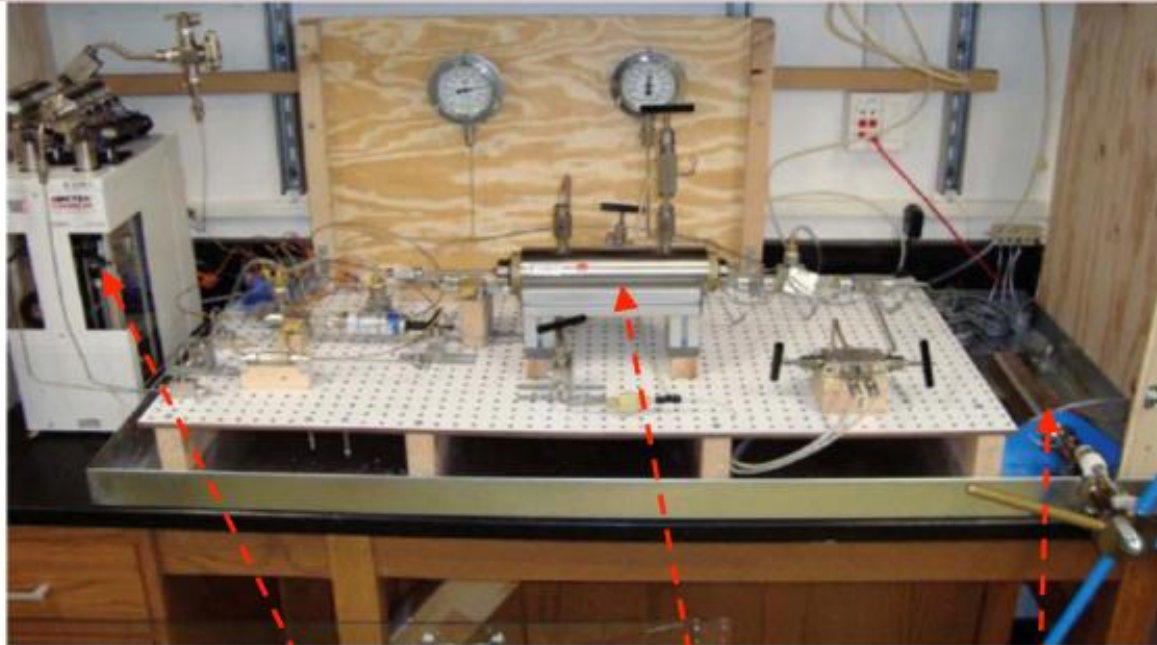
Technical Status

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- **Permeability measurements and application of the Klinkenberg effect**
- Molecular Dynamics simulations for permeability and viscosity estimates

Permeability, Effective Stress and Slip Flow

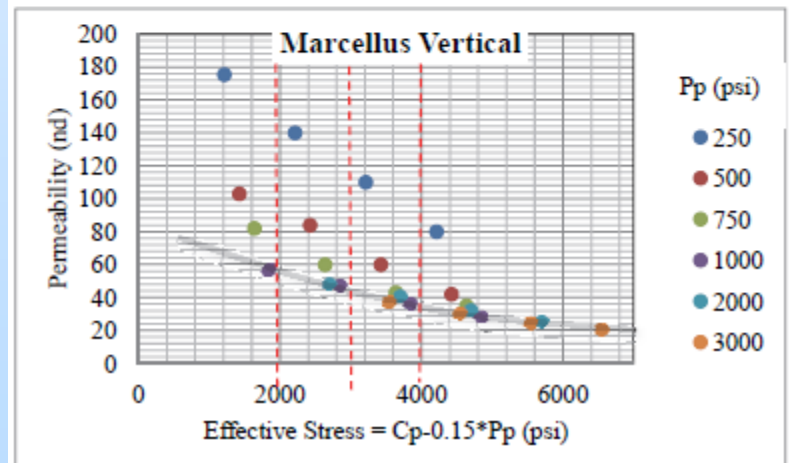
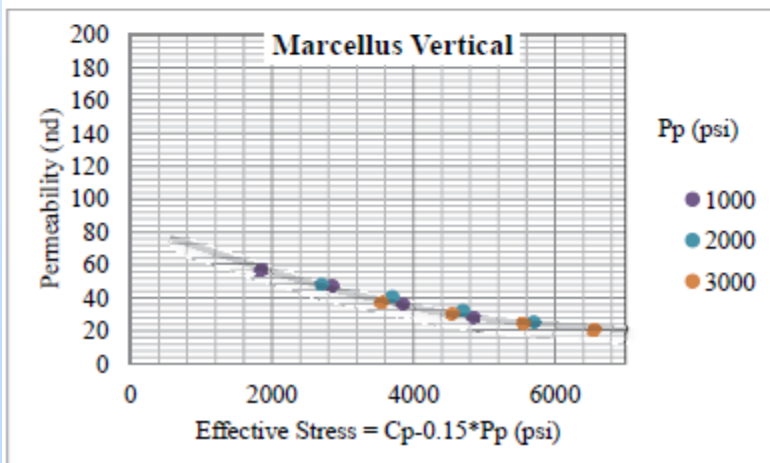
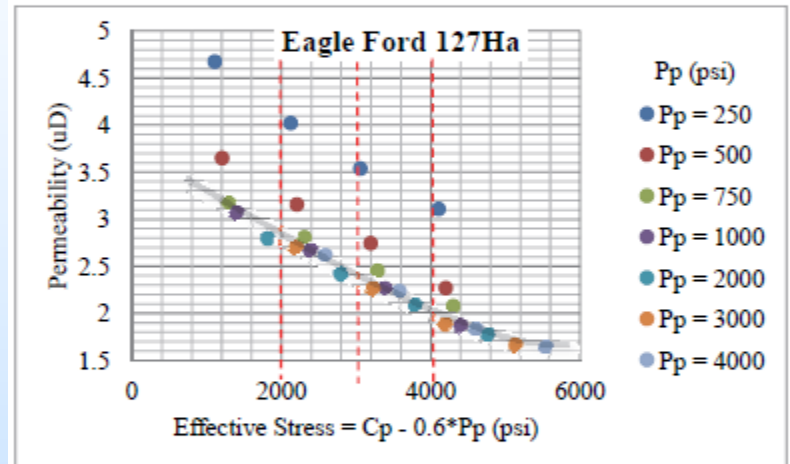
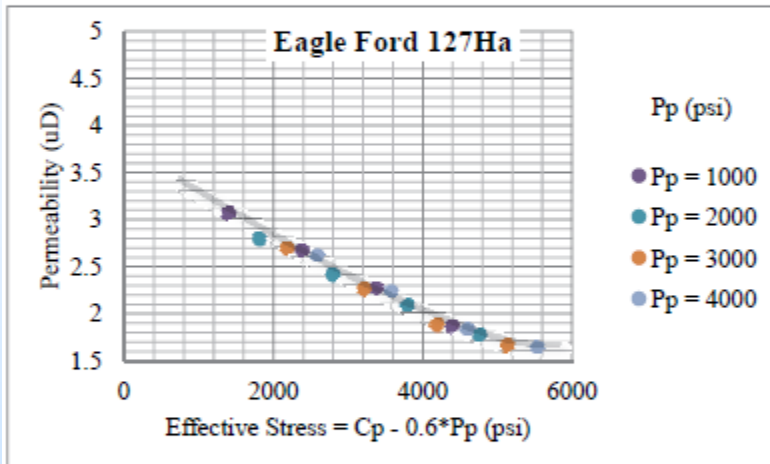
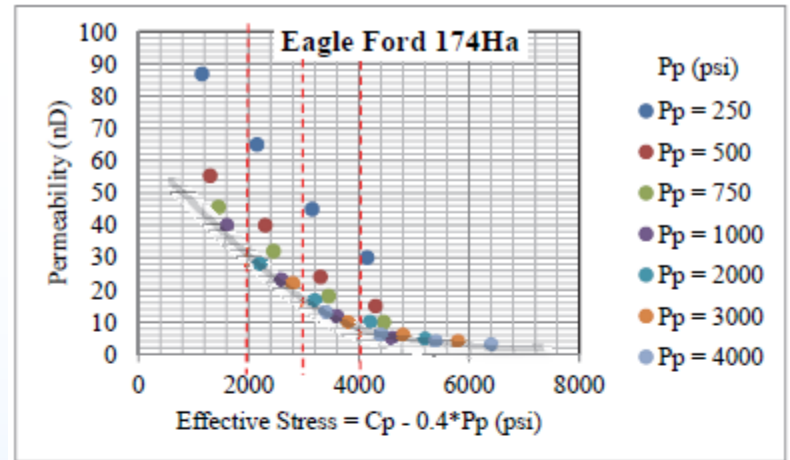
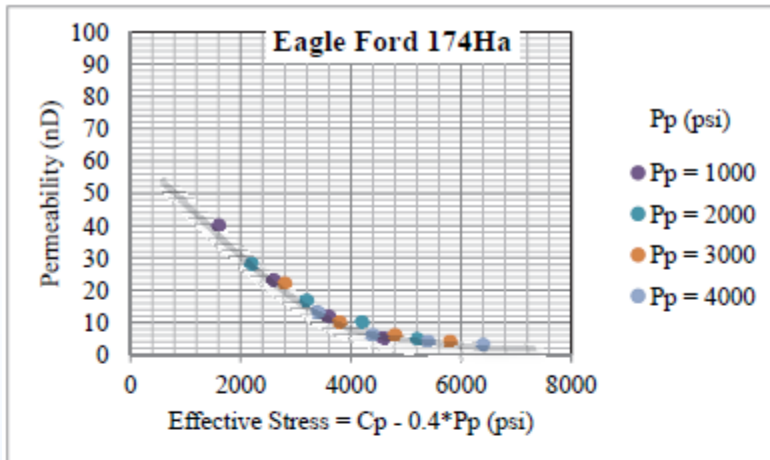
- Permeability varies with P_p and C_p due to:
 - Effective stress effects
 - Slippage effects
- Laboratory studies to date have neglected to account for both of these effects

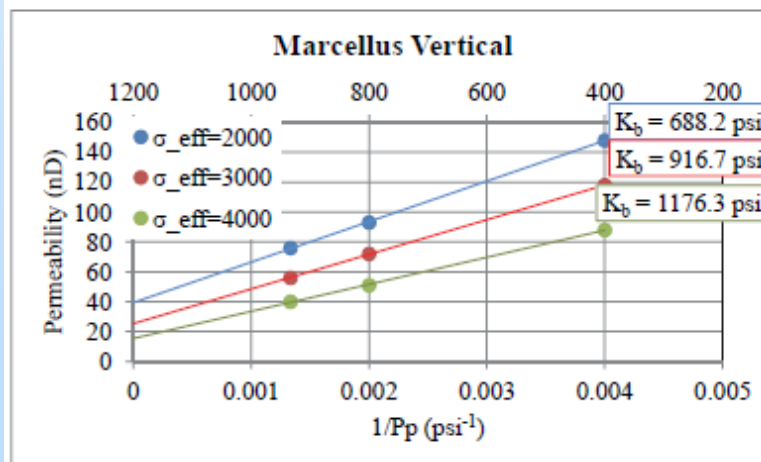
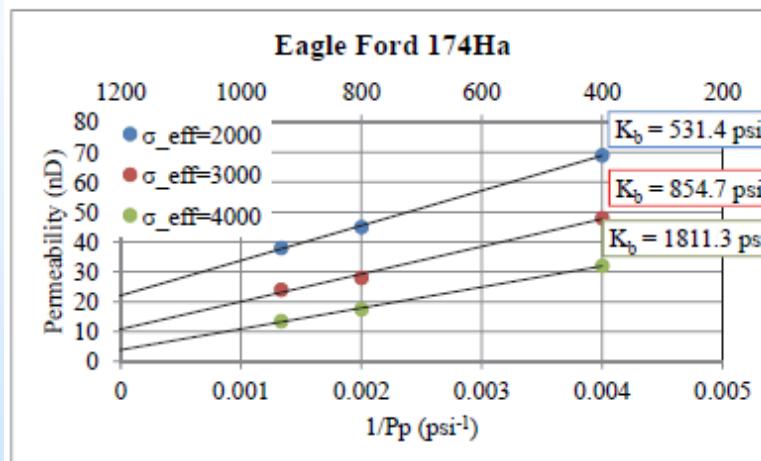
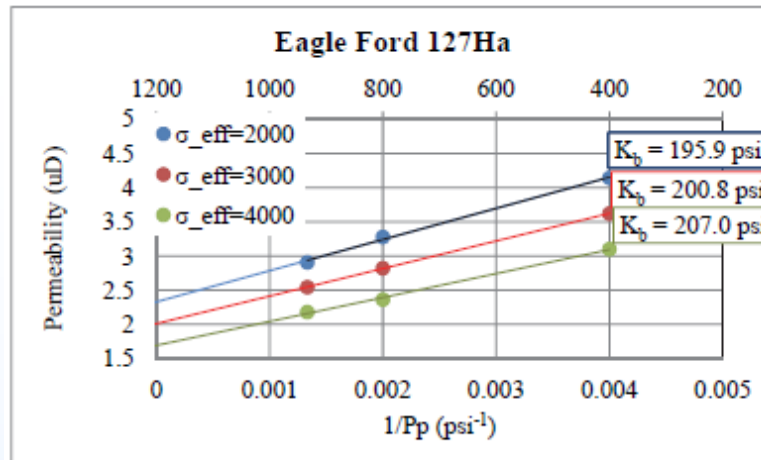
Permeability System Setup



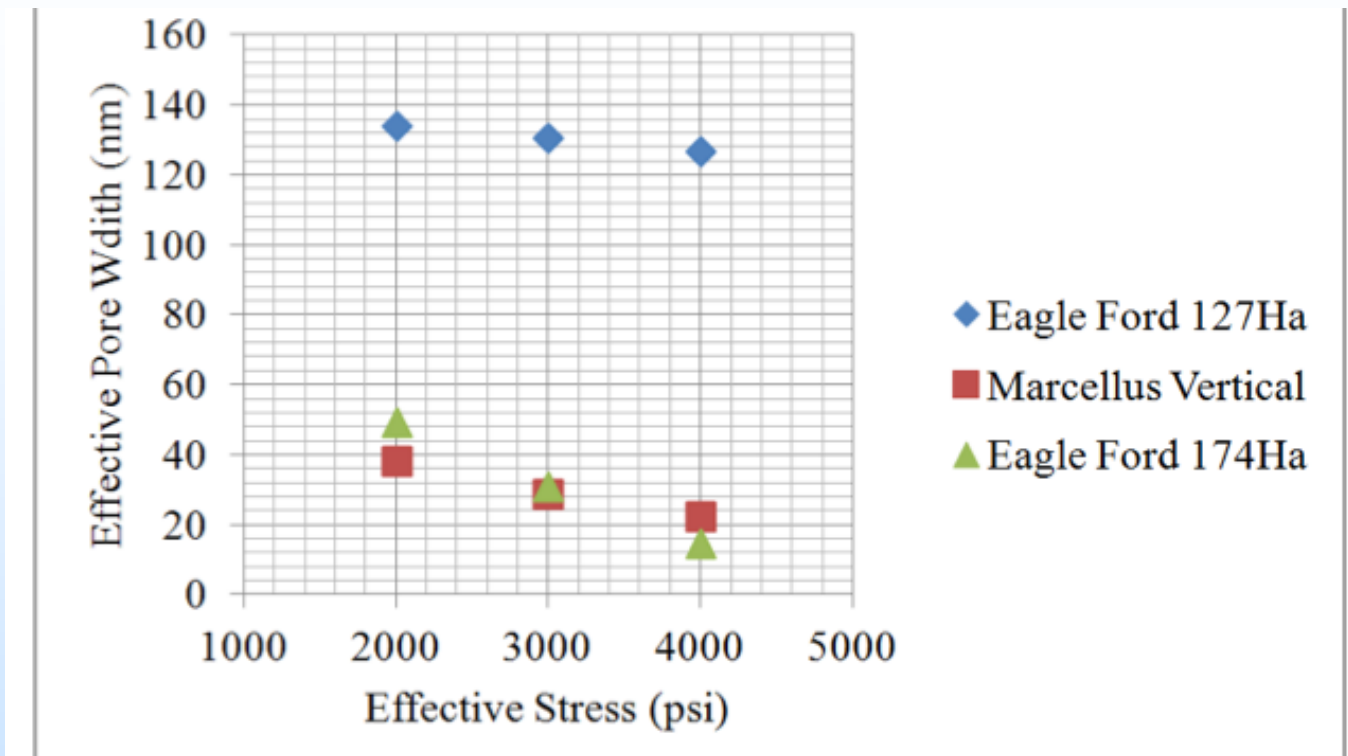
Summary of all Samples Measured

	Barnett 31 H -	60 – 160 nd
	Barnett 27 H -	800 – 1800 nd
	Haynesville G31 -	50 – 150 nd
	Montney H1 -	1 – 5 ud
	Eagle Ford 127Ha -	1.5 – 3.5 ud
	Marcellus Vertical -	20 – 180 nd
	Eagle Ford 174Ha -	5 – 90 nd





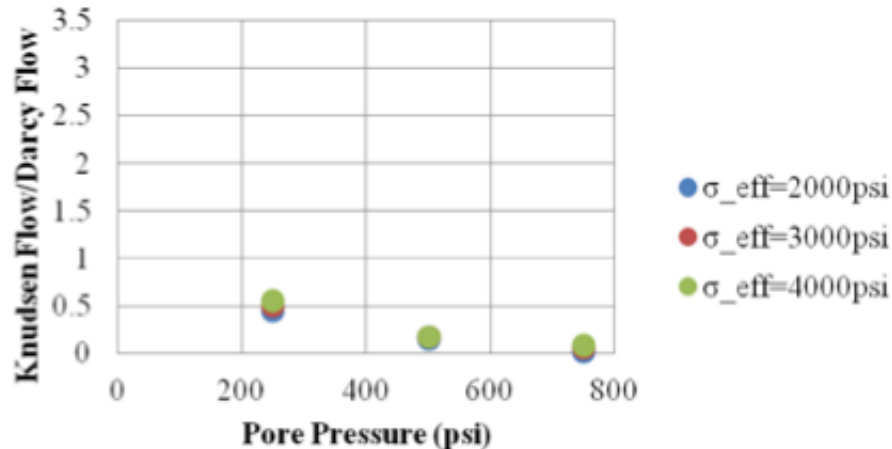
Effective Pore Size vs Effective Stress



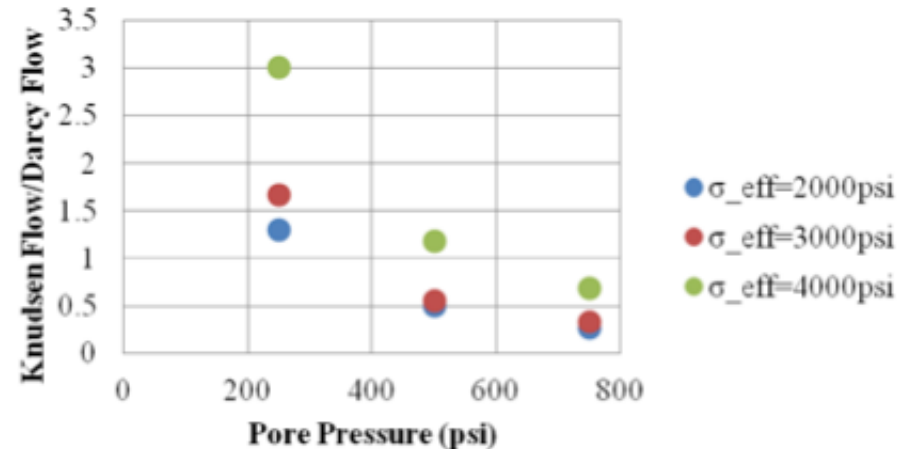
- Pore width decreases with increasing effective stress
- Pore widths range from 20 – 40 nm in Marcellus and Barnett samples, ~ 130 nm in Eagle Ford
- Klinkenberg pore sizes consistent with SEM images

To What Extent Does Knudsen Diffusion Contribute to Flow?

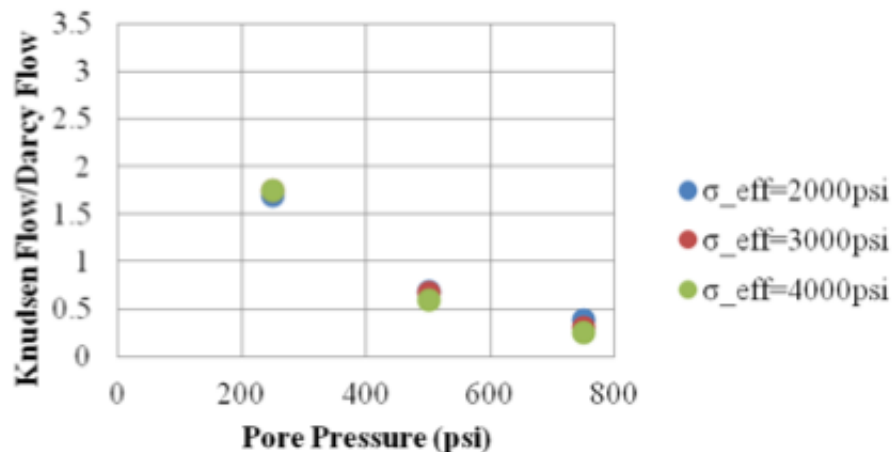
Eagle Ford 127Ha



Eagle Ford 174Ha



Marcellus Vertical

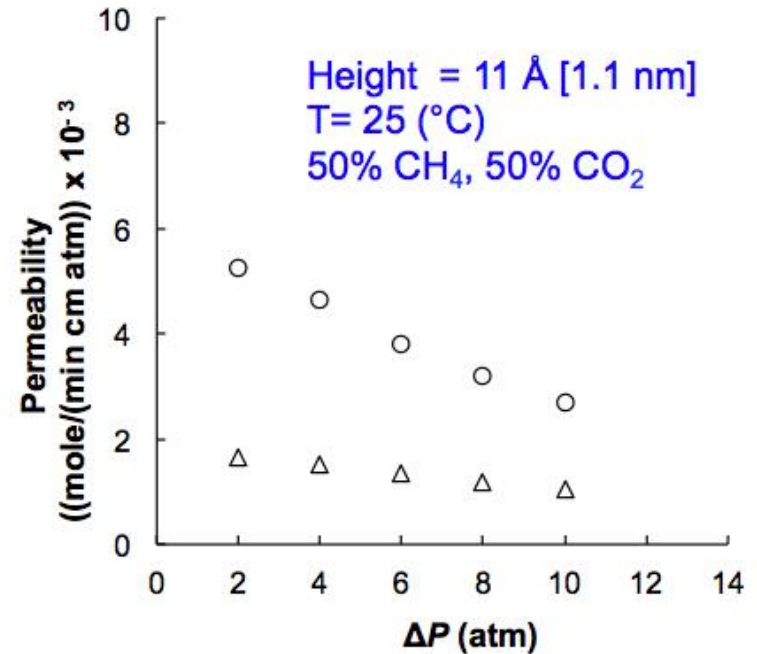
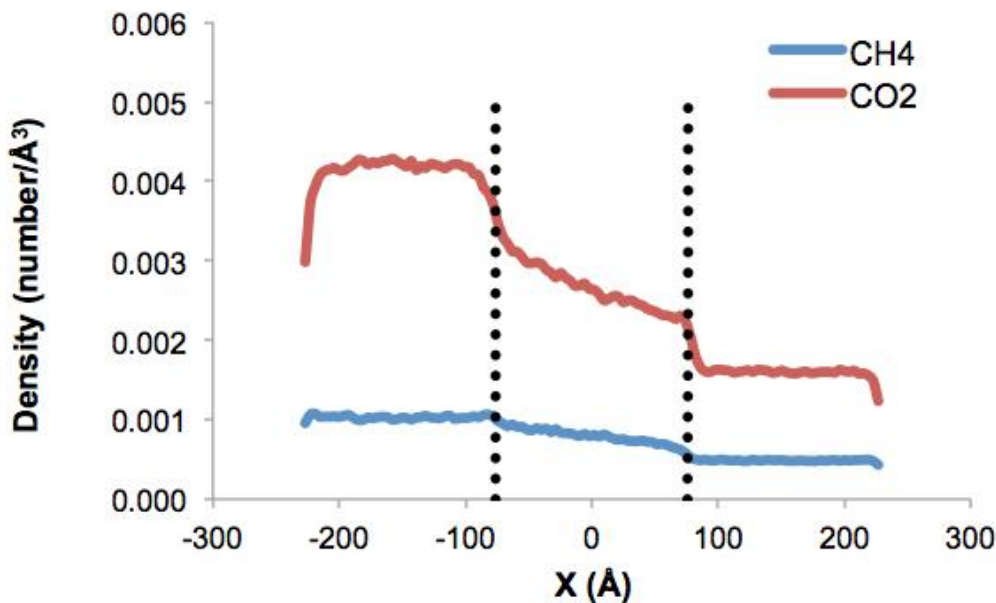
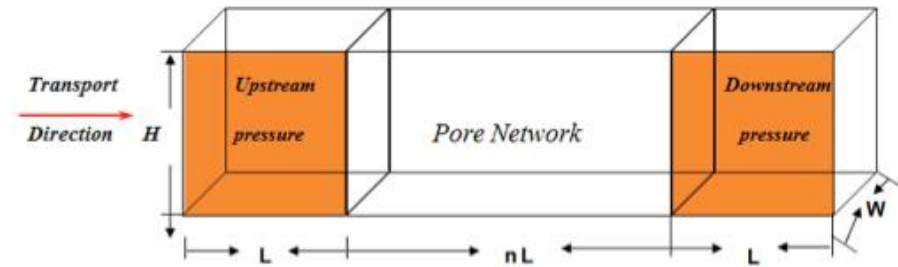
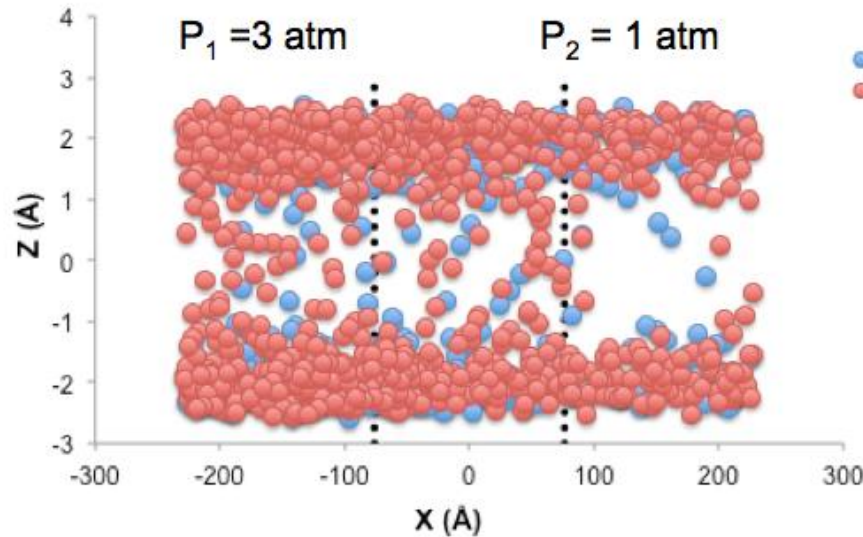


- Diffusive flow contributes appreciably to total flow at pore pressure <800 psi
- Diffusive flow is sometimes more important than Darcy flow at pore pressure <500 psi
- As we increase effective stress for a given pore pressure, we narrow the pore aperture and the relative contribution of diffusion increases

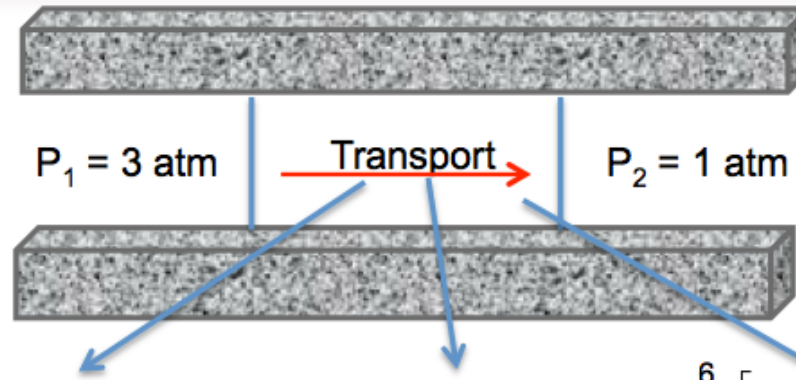
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Transport in a Slit Pore Model

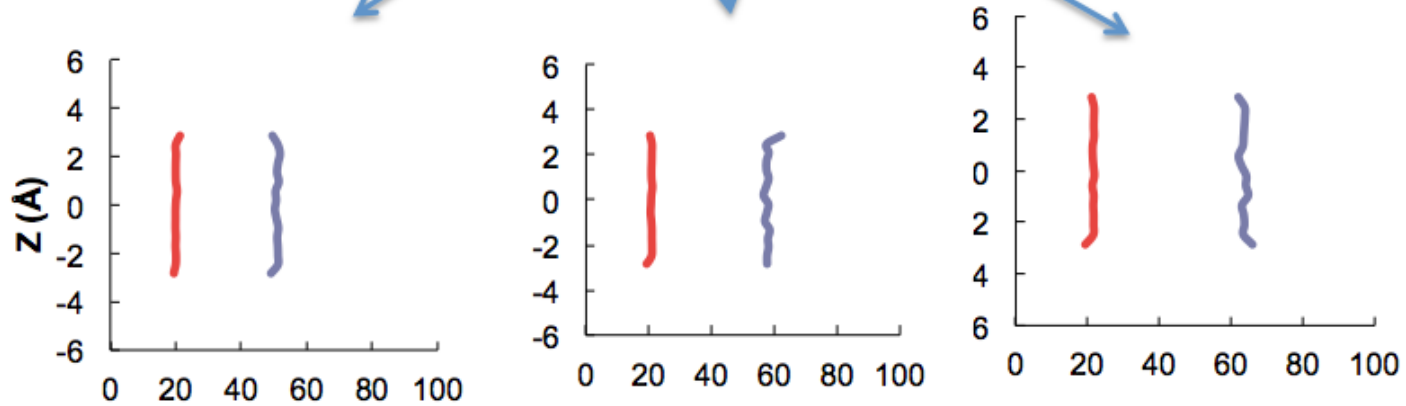


CH₄, CO₂ Velocity Profiles in Micropores (~ 1nm)

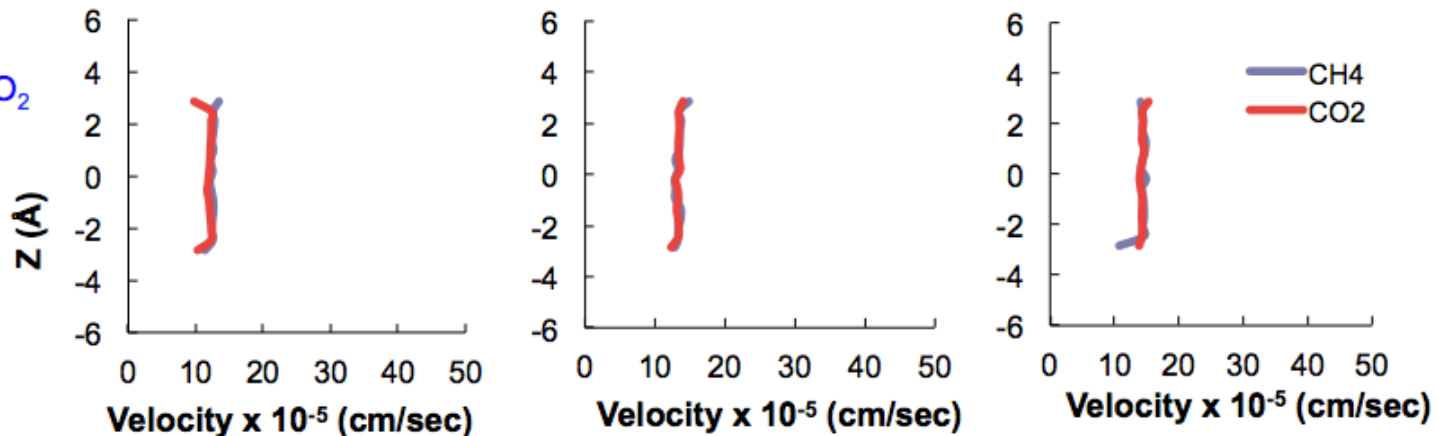


Length = 152 Å [15.2 nm]
 Width = 76 Å [7.6 nm]
 Height = 11 Å [1.1 nm]

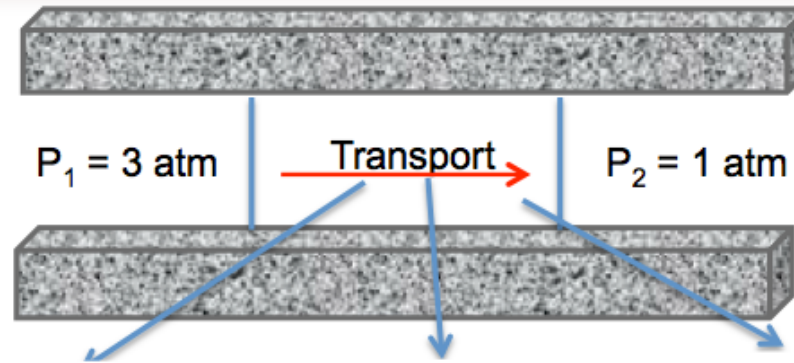
Pure CH₄, CO₂



50% CH₄, 50% CO₂

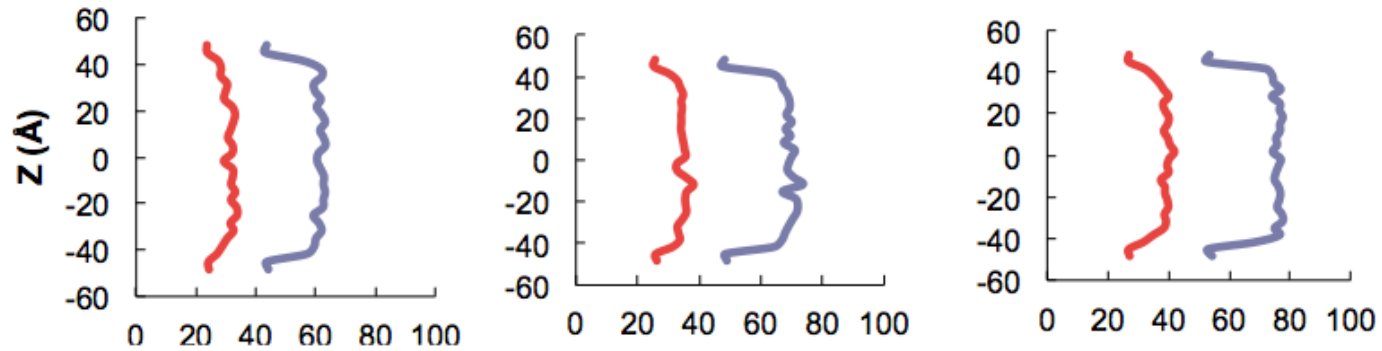


CH₄, CO₂ Velocity Profiles in Mesopores (10 nm)

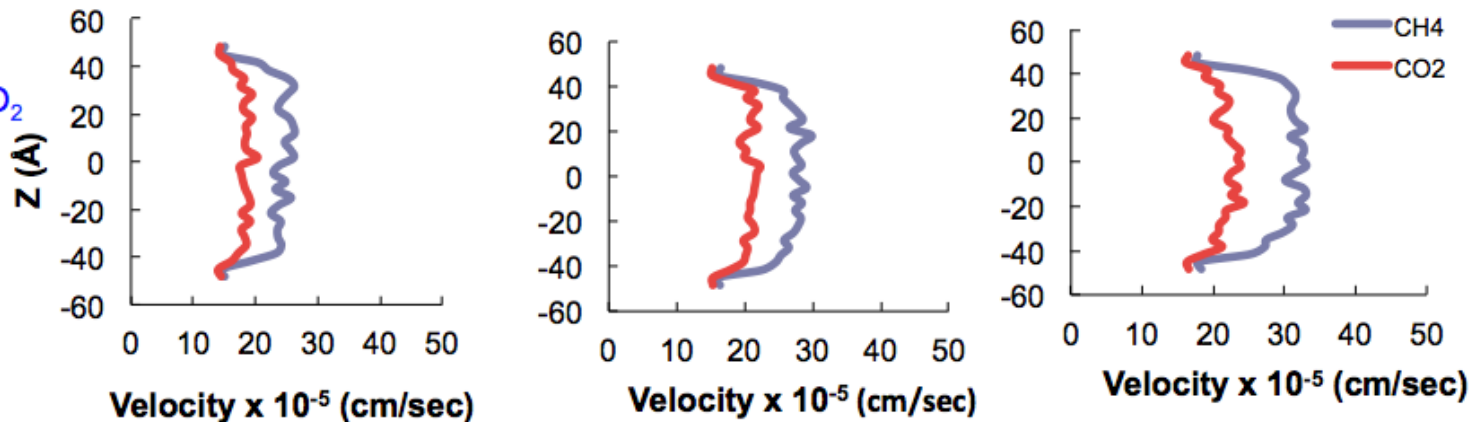


Length = 152 Å [15.2 nm]
 Width = 76 Å [7.6 nm]
 Height = 100 Å [10 nm]

Pure CH₄, CO₂



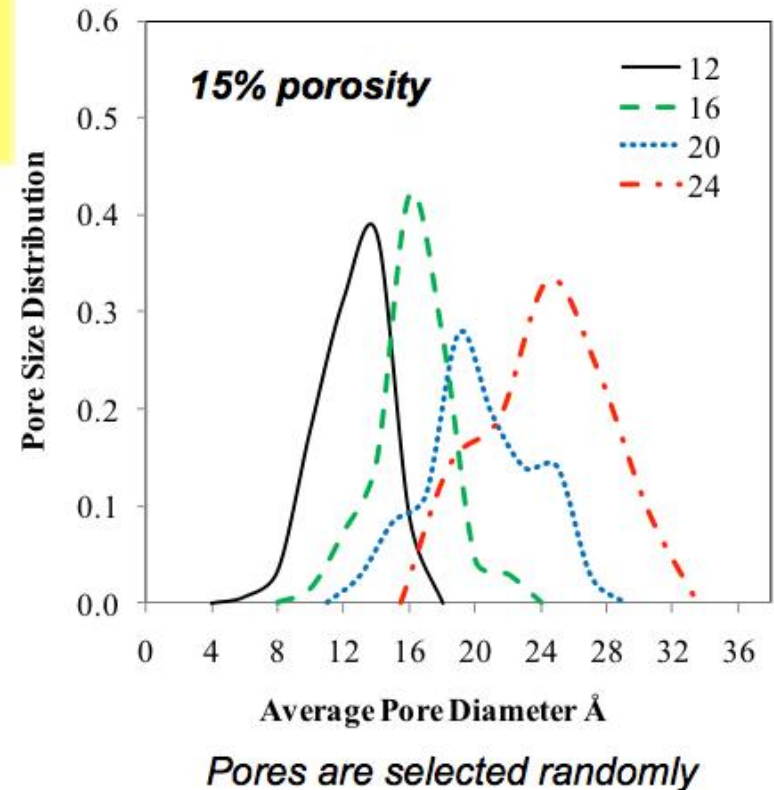
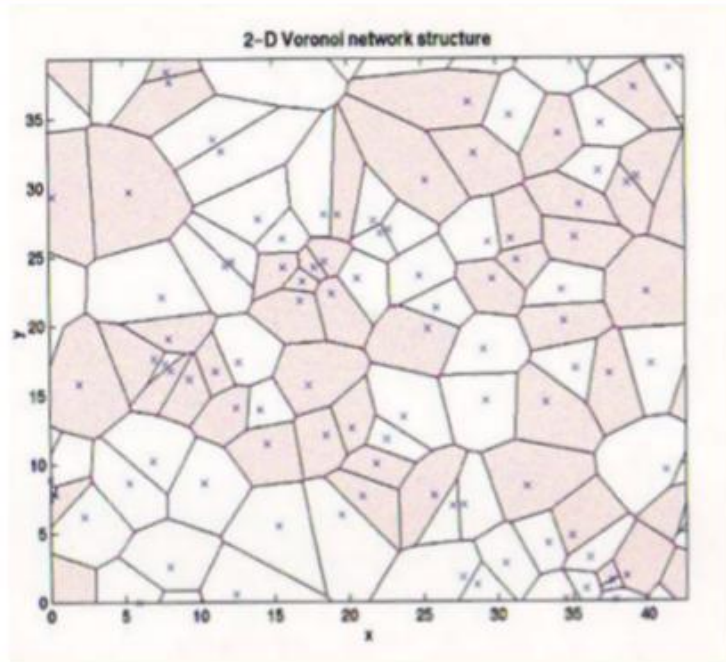
50% CH₄, 50% CO₂



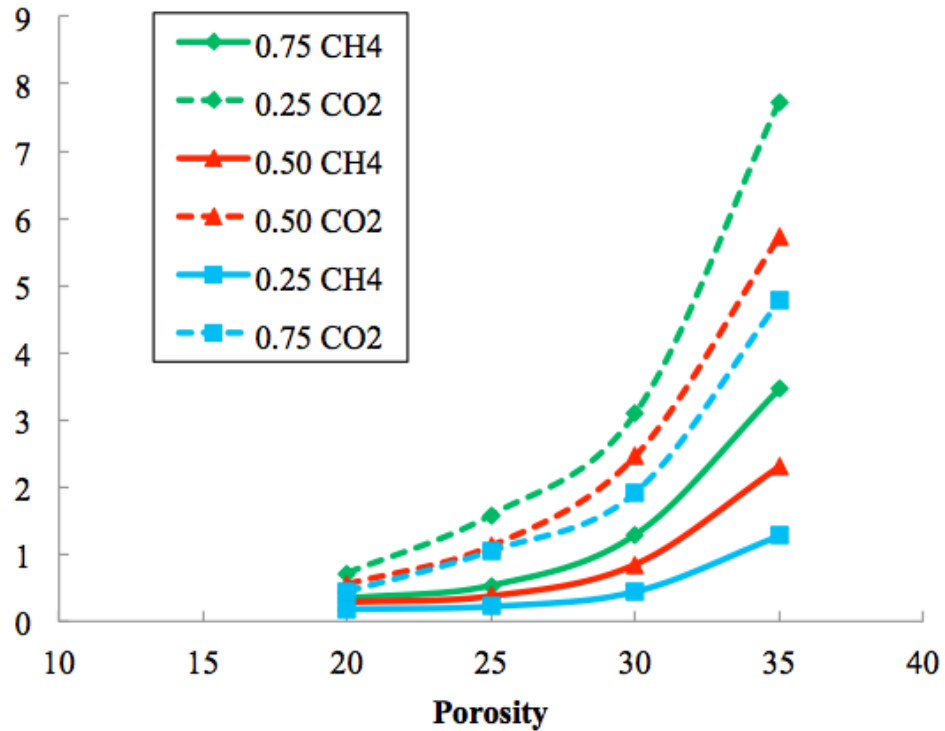
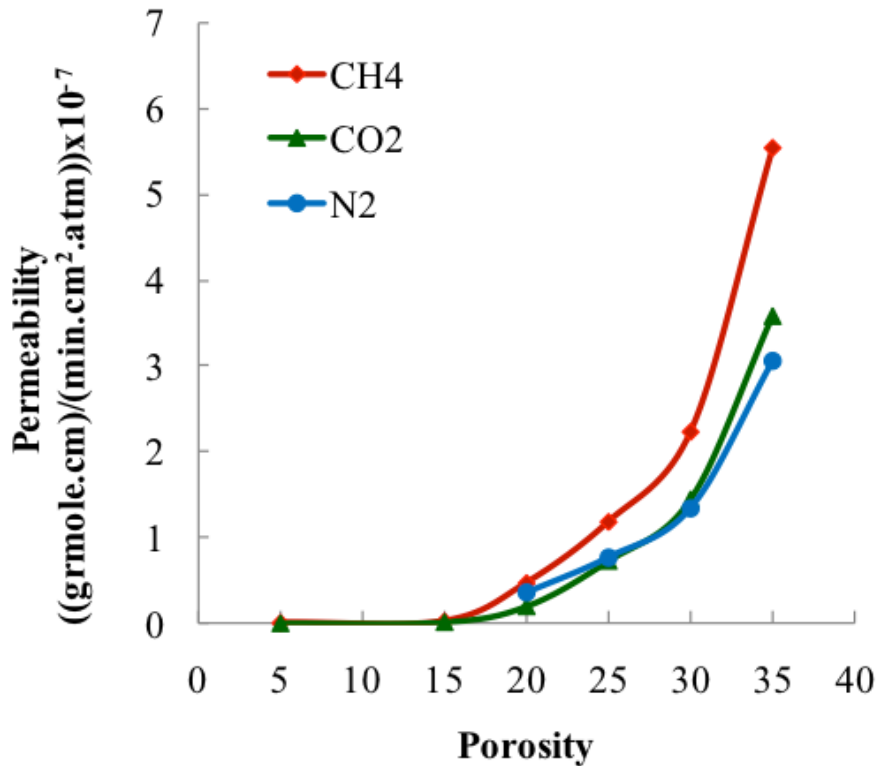
3D Pore Network Model

- 3D molecular pore network model based on the *Voronoi* tessellation method

First, we create a 3D simulation box of structural atoms corresponding to porous structure and then we tessellate the atomic structural box



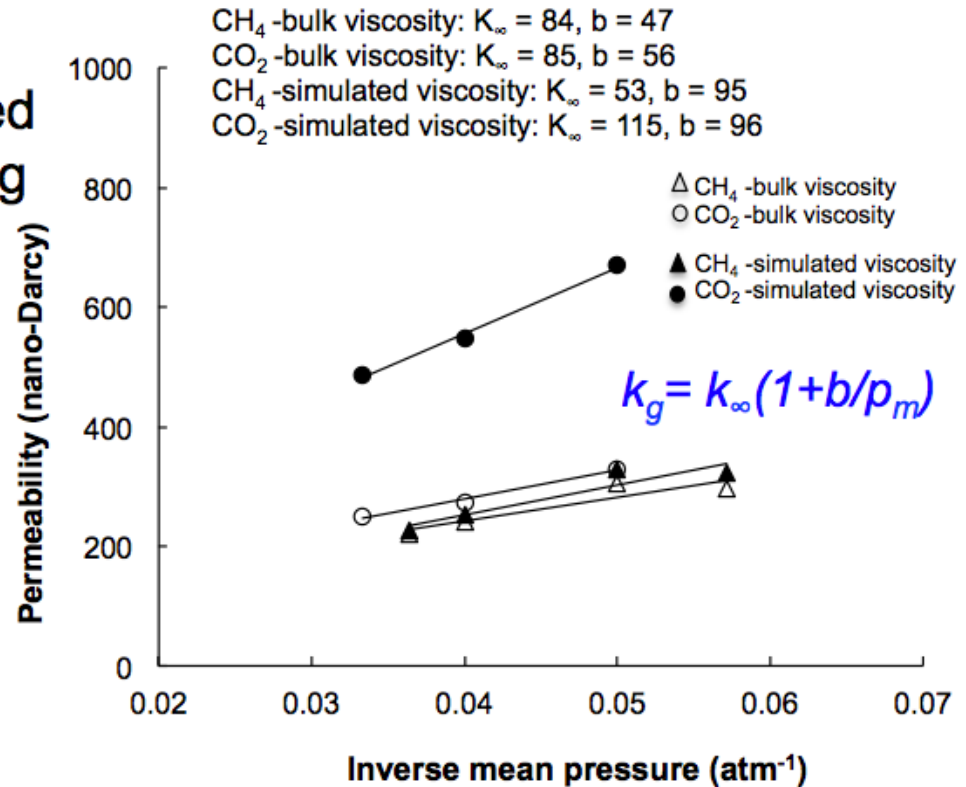
Permeability vs Porosity



*The average pore size in the 3-D pore network is 12 Å [1.2 nm]
 The upstream and downstream pressures are 50 and 20 atm, $T = 25$ (°C)
 LJ sizes of CH_4 , CO_2 and N_2 are 3.81, 3.794, and 3.694 Å respectively*

Viscosity Effect on Permeability

- The viscosity effect on the CO₂ permeabilities is more noticeable than CH₄, which results in increased permeability for CO₂ when reporting permeability in Darcy units using modeled viscosity
- The use of the bulk-phase CH₄ viscosity is a reasonable assumption as CH₄ is less influenced by the pore walls compared with CO₂



Green-Kubo and Einstein relations:

$$D = \frac{1}{3N} \sum_{i=1}^N \int_0^{\infty} \langle v_i(t) \cdot v_i(0) \rangle dt \quad \mu = \frac{k_B T}{3\pi d D}$$

*The average pore size in the 3-D pore network is 20 Å
 and the porosity is 20%
 The downstream pressure is fixed at 10 atm*

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Accomplishments to Date

- Furthered attempts to image in real time movement in situ of gas through shale cores in lab
- Imaged gas porosity of shale cores using X-ray CT scanning at mm and micron scales
- Investigated sorption, swelling and viscous creep in clays
- Designed and investigated model systems for simulating sorption and transport at micro and mesoscales
- Determined the extent of Knudsen diffusion on the transport mechanism at the nanoscale through application of the Klinkenberg effect
- Determined the difference in gas viscosity and density parameters from nano to micron scales

Summary

- Shale imaging across scales is required to understand the pore structure of the shale and the role the various pores play in gas transport and/or storage
- Gas slippage (Klinkenberg) can be used to determine the dominant transport mechanism at the nanoscale
- Future Plans
 - Continue with nano-scale imaging at SLAC
 - Continue with Klinkenberg investigations for adsorbing gas, CO₂
 - Compare scales and estimates of gas permeabilities of experiments vs theoretical predictions
 - Gas sorption will be used (Quantachrome Autosorb) to determine PSD of various shale samples

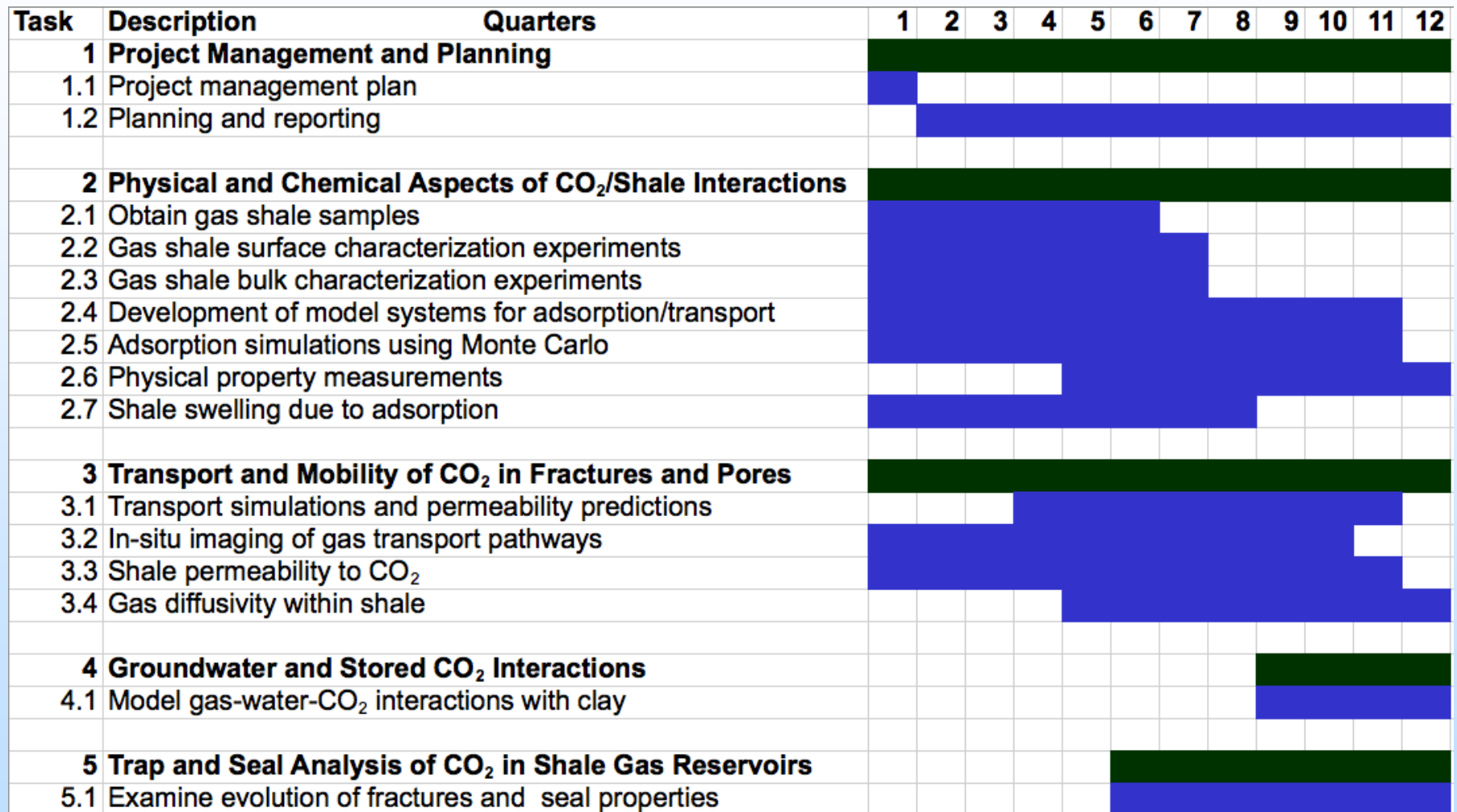
Appendix

- Organization
- Gantt Chart
- Bibliography

Organization

- Stanford University, School of Earth Sciences
 - PI: Professor Mark Zoback, Department of Geophysics
 - Dr. Sander Hol (Post-doc) and Rob Heller (PhD student)
 - Co-PI: Professor Tony Kocscek, Energy Resources Engineering Department
 - Bolivia Vega (Research Assistant), Dr. Cindy Ross (Research Associate) and Khalid Alnoaimi (PhD student)
 - Co-PI: Assistant Professor Jennifer Wilcox, Energy Resources Engineering Department
 - Dr. Mahnaz Firouzi (Post-doc), Dr. Dawn Geatches (Post-doc), and Dr. Yangyang Liu (graduated w/ PhD in June 2012)

Gantt Chart



Bibliography

- Liu, Y. and Wilcox, J. Effects of Surface Heterogeneity on the Adsorption of CO₂ in Microporous Carbons, *Environmental Science and Technology*, 46(3), p. 1940, 2012.
- Liu, Y. and Wilcox, J. CO₂ Adsorption on Carbon Models of Organic Constituents of Gas Shale and Coal, *Environmental Science and Technology*, 45(2), p. 809, 2011.
- Firouzi, M. and Wilcox, J., Molecular Modeling of CO₂ Transport and Storage in Porous Carbon-based Materials, *J. Micro and Mesoporous Materials*, 158, p. 195, 2012.
- Firouzi, M. and Wilcox, J., Slippage and Viscosity Predictions in Carbon Micropores from Molecular Simulation and Influence on CO₂ and CH₄ Transport, *Physics Review E*, in review.